

## NUMERICAL PROBLEMS

18.1 The length of a space ship is measured to be exactly one-third of its proper length. What is the speed of the spaceship relative to the observer?

**Solution**

Proper length =  $L_0$

Measured length  $L = \frac{L_0}{3}$   $v = ?$

Formula for Lorentz length contraction is

$$L = L_0 \sqrt{1 - \left(\frac{v^2}{c^2}\right)}$$

According to given condition ( $L = \frac{L_0}{3}$ ) above equation can be written as

$$\frac{L_0}{3} = L_0 \sqrt{1 - \left(\frac{v^2}{c^2}\right)}$$

$\Rightarrow$

$$\frac{1}{3} = \sqrt{1 - \left(\frac{v^2}{c^2}\right)} \quad \text{Squaring both sides \& solving for}$$

$v$

$\Rightarrow$

$$\frac{1}{9} = 1 - \left(\frac{v^2}{c^2}\right) \Rightarrow \frac{v^2}{c^2} = \frac{8}{9}$$

$$v = 0.9428 c$$

18.2 The time period of a pendulum is measured to be 3s in inertial frame of the pendulum. What is the period when measured by an observer moving with a speed of  $0.95c$  with respect to the pendulum?

**Solution**

$t_0 = 3$  sec

$t = ?$  If speed  $v = 0.95 c$

Equation for time dilation is

$$t = \frac{t_0}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}} = \frac{3}{\sqrt{1 - \left(\frac{(0.95c)^2}{c^2}\right)}} = \frac{3}{\sqrt{1 - (0.9025)}}$$

$$t = 9.6 \text{ sec}$$

**18.3 An electron, which has a mass  $9.11 \times 10^{-31}$  kg, moves with a speed of  $0.75c$ . Find its relativistic momentum and compare this value with the momentum calculated from classical expression.**

**Solution**

$m_e = 9.11 \times 10^{-31}$  kg       $v = 0.75 c$

Relativistic momentum  $P_r = ?$  and compare with  $P_{\text{classical}}$

$$P_r = m v = \frac{m_0 v}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$$

$$\frac{P_r}{P_{\text{classical}}} = 1.51$$

$$P_r = 1.51 P_{\text{classical}}$$

$$P_r = \frac{(9.11 \times 10^{-31})(0.75 \times 3 \times 10^8)}{\sqrt{1 - \left(\frac{0.75 c}{c}\right)^2}}$$

$$P_r = 3.1 \times 10^{-22} \text{ kg m s}^{-1}$$

$$P_{\text{classical}} = m_0 v$$

$$P_{\text{classical}} = (9.11 \times 10^{-31})(0.75 \times 3 \times 10^8)$$

$$P_{\text{classical}} = 2.05 \times 10^{-22} \text{ kg m s}^{-1}$$

For comparison, we find % difference as  $\left(\frac{P_r - P_{\text{classical}}}{P_{\text{classical}}}\right) \times 100\% = 51\%$

**18.4 An electron moves with a speed of  $v = 0.85c$ . Find its total energy and K.E in electron volt.**

**Solution**

$v = 0.85c$

T.E = ? (in eV)

E = ? (in eV)

At the given speed ( $0.85c$ ) the relativistic mass  $m$  is

$$m = \frac{m_0}{\sqrt{1 - \left(\frac{v}{c}\right)^2}} = \frac{9.11 \times 10^{-31}}{\sqrt{1 - \left(\frac{0.85 c}{c}\right)^2}} = 1.73 \times 10^{-30} \text{ kg}$$

Now Total energy T.E is

$$T.E = mc^2 = (1.73 \times 10^{-30})(3 \times 10^8)^2$$

$$T.E = 1.56 \times 10^{-13} \text{ J}$$

divide by  $1.6 \times 10^{-19} \text{ J / eV}$

$$T.E = 0.973 \text{ MeV}$$

Now total energy T.E and kinetic energy K.E are related to rest energy  $E_0$

$$T.E = K.E + E_0$$

$$K.E = T.E - E_0$$

$$K.E = (0.973 - 0.5124) \text{ MeV}$$

$$K.E = 0.461 \text{ MeV}$$

$$E_0 = m_0 c^2 = (9.11 \times 10^{-31})(3 \times 10^8)^2$$

$$E_0 = 8.199 \times 10^{-17} \text{ J}$$

divide by "e"

$$E_0 = 0.5124 \text{ MeV}$$

**18.5 Rest mass of a proton is  $1.67 \times 10^{-27}$  kg. At what speed would the mass of the proton be tripled?**

**Solution**

$m_0 = 1.67 \times 10^{-27}$  kg       $v = ?$  at which  $m = 3m_0$

Relativistic mass of proton is given by

$$m = \frac{m_0}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}}$$

According to the given condition ( $m = 3m_0$ )

$$3m_0 = \frac{m_0}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}}$$

taking reciprocal

$$\frac{1}{9} = 1 - \left(\frac{v^2}{c^2}\right) \Rightarrow \frac{v^2}{c^2} = \frac{8}{9}$$

Taking square root

$$v = 0.9428c$$

**18.6 At what fraction of speed of light must a particle move so that its K.E is one and a half times its rest energy?**

Solution

$$V = ?$$

when  $K.E = \frac{3}{2} E_0 = \frac{3}{2} (m_0 c^2)$

$E =$  kinetic energy K.E + rest energy  $E_0$

$$mc^2 = \frac{3}{2} (m_0 c^2) + (m_0 c^2) \quad \text{divide by } c^2$$

$$m = \frac{3}{2} m_0 + m_0 = \frac{5}{2} m_0$$

$$\frac{m_0}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}} = \frac{5}{2} m_0 \quad \text{As } m = \frac{m_0}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}}$$

$$\frac{1}{\sqrt{1 - \left(\frac{v^2}{c^2}\right)}} = \frac{5}{2}$$

Take reciprocal and square

$$1 - \left(\frac{v^2}{c^2}\right) = \frac{4}{25}$$

$$\frac{v^2}{c^2} = \frac{21}{25}$$

Taking square root & solving for "v"

$$v = 0.916c$$

**18.7 A metal, whose work function is 3.0 eV, is illuminated by light of wavelength  $3 \times 10^{-7} \text{m}$ . Calculate (a) The threshold frequency, (b) The maximum energy of photoelectrons (c) The stopping potential.**

Solution

$$\Phi = 3 \text{eV} = 3 \times 1.6 \times 10^{-19} = 4.8 \times 10^{-19} \text{ J}$$

$$\lambda = 3 \times 10^{-7} \text{m} \quad \text{(a) } f_0 = ?$$

$$\text{(b) } K.E_{\text{max}} = ?$$

$$\text{(c) } V_0 = ?$$

(a) Work function is the minimum amount of energy required for ejecting electrons from a metal surface

$$\Phi = h f_0$$

$$f_0 = \frac{\phi}{h} = \frac{4.8 \times 10^{-19}}{(6.63 \times 10^{-34})} = 0.72 \times 10^{15} \text{ Hz}$$

(b) According to the Einstein's equation of photoelectric effect

$$E = \Phi + K.E_{\text{max}}$$

$$K.E_{\text{max}} = hf - \phi$$

Since

$$f = \frac{c}{\lambda} = \frac{3 \times 10^8}{(3 \times 10^{-7})} = 10^{15} \text{ Hz}$$

$$K.E_{\text{max}} = (6.63 \times 10^{-34}) (10^{15}) - (4.8 \times 10^{-19})$$

$$K.E_{\text{max}} = 1.83 \times 10^{-19} \text{ J} \quad \text{divide by } 1.6 \times 10^{-19}$$

$$K.E_{\text{max}} = 1.14 \text{ eV}$$

(c)

$$K.E_{\text{max}} = eV_0 \quad \text{where } V_0 \text{ is stopping potential}$$

⇒

$$V_0 = \frac{K.E_{\text{max}}}{e} = \frac{1.14 \times 1.6 \times 10^{-19}}{1.6 \times 10^{-19}}$$

⇒

$$V_0 = 1.14 \text{ V}$$

**18.8 The thermal radiation from the sun peaks in the visible part of the spectrum.**

**Estimate the temperature of the sun.**

**Solution**

$T = ?$  For visible part of spectrum

For visible part of spectrum, wavelength is 500 nm

$$\lambda_{\text{max}} = 500 \text{ nm}$$

Using Wien's displacement law

$$\lambda_{\text{max}} T = 0.2898 \times 10^{-2} \text{ m K}$$

$$T = \frac{0.2898 \times 10^{-2}}{\lambda_{\text{max}}} = \frac{0.2898 \times 10^{-2}}{500 \times 10^{-9}}$$

$$T = 5796 \text{ K} \approx 5800 \text{ K}$$

**18.9 A 50 keV X-ray is scattered through an angle of  $90^\circ$ . What is the energy of X-ray after Compton scattering?**

**Solution**

Initially  $E = 50 \text{ keV} = 50,000 \times 1.6 \times 10^{-19} \text{ J} = 8 \times 10^{-15} \text{ J}$

$\theta = 90^\circ$  Energy after scattering  $E' = ?$

Energy of X-ray photon after scattering can be found by equation

$$E' = \frac{hc}{\lambda'} \quad (1)$$

$\lambda'$  is wavelength of scattered photon and is given by the equation

$$\lambda' = \lambda + \frac{h}{m_0 c} (1 - \cos\theta) \quad (2)$$

$\lambda$  is the only unknown at the right hand side for which we use

$$E = \frac{hc}{\lambda}$$

$$\lambda = \frac{hc}{E} = \frac{(6.63 \times 10^{-34})(3 \times 10^8)}{8 \times 10^{-15}}$$

$\lambda = 2.49 \times 10^{-11} \text{ m}$  putting this value in eq (2), we get;

$$\lambda' = 2.49 \times 10^{-11} + \frac{6.63 \times 10^{-34}}{(9.11 \times 10^{-31})(3 \times 10^8)} (1 - \cos 90^\circ)$$

$$\lambda' = 2.73 \times 10^{-11} \text{ m} \quad \text{put this in eq (1) we get}$$

$$E' = \frac{(6.63 \times 10^{-34})(3 \times 10^8)}{2.73 \times 10^{-11}}$$

$$E' = 7.28 \times 10^{-15} \text{ J}$$

$$E' = 45.5 \text{ keV} \quad \text{divide by "e"}$$

**18.10 Calculate the wavelength of de Broglie waves associated with electrons accelerated through a potential difference of 200 V.**

**Solution**  
 $\lambda = ?$  (for electron)

$$V_0 = 200 \text{ V}$$

De-Broglie's wavelength associated with a moving electron is given by

$$\lambda = \frac{h}{mv} \quad (1)$$

$$\text{Kinetic energy } \frac{1}{2} m v^2 = e V_0$$

$$v = \sqrt{\frac{2 e V_0}{m}} = \sqrt{\frac{2 (1.6 \times 10^{-19}) (200)}{9.11 \times 10^{-31}}}$$

$$v = 8.38 \times 10^6 \text{ m/s}$$

$$\text{Put this value in eq (1) we get } \lambda = \frac{6.63 \times 10^{-34}}{(9.11 \times 10^{-31})(8.38 \times 10^6)} = 0.87 \text{ \AA}$$

**18.11 An electron is accelerated through a potential difference of 50V. Calculate its de Broglie Wavelength.**

$$V_0 = 50 \text{ V}$$

$\lambda = ?$  (for electron)

**Solution**

De-Broglie's wavelength associated with a moving electron is given by

$$\lambda = \frac{h}{mv} \quad (1)$$

$$\frac{1}{2} m v^2 = e V_0$$

Re-arranging them we get

$$v = \sqrt{\frac{2 e V_0}{m}}$$

$$v = \sqrt{\frac{2 (1.6 \times 10^{-19}) (50)}{9.11 \times 10^{-31}}} = 4.19 \times 10^6 \text{ m/s}$$

$$\lambda = \frac{6.63 \times 10^{-34}}{(9.11 \times 10^{-31})(4.19 \times 10^6)} = 1.74 \text{ \AA}$$

Put this value in eq (1) we get

**18.12 The speed of an electron is measured to be  $5 \times 10^3 \text{ m/s}$  to an accuracy of 0.003%. Find the uncertainty in determining the position of this electron.**

**Solution**

$$v = 5 \times 10^3 \text{ m/s}$$

$$\text{Accuracy } A_c = 0.003\% = \frac{0.003}{100} = 3 \times 10^{-5}$$

Uncertainty in position  $\Delta x = ?$  ( for electron )

Uncertainty in velocity  $\Delta v = 5 \times 10^3 \times A_c$

$$\Delta v = 5 \times 10^3 \times (3 \times 10^{-5}) = 0.15 \text{ m/s}$$

Now according to Heisenberg's uncertainty principle,

$$\Delta x \Delta P = h \quad (\text{As } \Delta P = m \Delta v)$$

$$\Delta x = \frac{h}{m \Delta v} = \frac{6.63 \times 10^{-34}}{(9.11 \times 10^{-31}) (0.15)} = 4.85 \times 10^{-3} \text{ m}$$

**18.13 The life time of an electron in an excited state is about  $10^{-8}$  sec. What is its uncertainty in energy during this time?**

**Solution**

$$\Delta t = 10^{-8} \text{ sec} \quad \Delta E = ?$$

By Heisenberg's uncertainty principle,

$$\Delta E \Delta t = h$$

$$\Delta E = \frac{h}{\Delta t} = \frac{6.63 \times 10^{-34}}{10^{-8}} = 6.63 \times 10^{-26} \text{ J}$$